76-32-5-29/47

Chemical Proof of the Formation of ${\rm H_2O_2}$ on the Atmospheric Corrosion of Metals

Member of the Academy of Sciences, as well as to the participants of the seminary. From the experimental part can be seen that two micro-methods were used for the determination of hydrogen peroxide vapors, i. e. one with the reagent being on the dried filter paper, and the other with the correspondingly prepared photofilm. The obtained results carried out by means of 12 reagents on 19 metals are mentioned in a table. The reagent by Deniges (Ref 29) proved to be most sensitive, as well as $K_4/Fe(CN_6)/FeCl_3$, which was also used on the

photographic film and for experiments on the kinetics of the formation of hydrogen peroxide on the atmospheric corrosion of magnesium. Lead sulfide was used analogously, however, not with quite as good results. The preparation of a Ni₂O₃--reagent is described as well. The formation of H₂O₂ was also observed on a stainless steel 18/8 and a gold alloy 916. Finally the author states that on the surface of electronegative, as well as electropositive metals hydrogen peroxide forms in the presence of humidity on ordinary atmospheric corrosion. The author thanks Ye. M. Vasil'yeva for her collaboration. There are 1 table and 32 references, 13 of which

Card 2/3

76-32-5-29/47

Chemical Proof of the Formation of $\mathrm{H_20_2}$ on the Atmospheric Corrosion of Metals

are Soviet.

ASSOCIATION: Odesskiy tekhnologicheskiy institut

(Odessa Technological Institute)

SUBMITTED: February 8, 1957

1. Hydrogen peroxide—Synthesis 2. Metals—Corrosion

3. Hydrogen peroxide vapors -- Determination

4. Photographic emulsions---Photochemical reactions

Card 3/3

· AUTHORS: Royth, I. L., Bolotich, I. P. SOV/ 20-120-1-31/63

TITLE: The Mechanism of the Vertical Distribution of H202 Over

Solution (Mekhanism vertikal nogo raspredeleniya ${\rm H_20_2}$ ned rastvorom)

PERIODICAL: Do'lady Abademii mout SSSR, 1930, Val. 120, Nr 1, 21. 115 - 118 (USSR)

ADSTRACT: In a provious paper by I.I. Royth (Ref 1) it was shown that the photographically active particles separating in the atmospheric correction at the mirfuge of metals decrease according to an or possential law with increasing distance. These particles are $\rm H_2O_2-$ notecules. It therefore may be expected that there also

is a vertical distribution for the hydrogen peroxide vapors so portion, from the solution. In order to prove this the following experiments were carried out: An agreeus solution of hydrogen peroxide of a certain concentration was powed into a bulb and a photographic plate was nameted at a small angle to the surface of the colution. The esperi ental conditions are discussed. A

linear dependence of the blac oning D on its height he above Card 1/4

The Mac amico of the Vertical Distribution of H₂O₂ 301/20-120-1-51/63 Over a Solution

the surface of the hydroten peroxide solution was found as a result of these experiments: D = -bh + h; where b and A denote constants. The obtained curves result from the averaging of the data of to experiments with constant duration of exposure. The in of the further inventiation on the salamation of the resolution of the vapors separated from the H₂O₂ solution. The number of hydroten permits aclesules de-

creases emponentially with the height above the solution. Also by theoretical considerations an equation is found which is analogous to the first written down and emperimentally determined on tion. The aloge of the particles depends on the number of the episting dust particles. With increasing number of the dust particles causing the decomposition of H_2O_2 the aloge of the straight decreases towards the axis of the alocises. In order to rove experimentally the mentioned formula the dependence of D on h in the case of dusty hir and hir partly free of dust where the surface of the H_2O_2 -solution was determined. The perfer use of the experiments is described. The results of these

Card 2/4

The Mechanism of the Vertical Distribution of H₂O₂ 30V/20-120-1-31/63

experiments tend to show the following: The decomposition of the hydrogen peroxide molecules evaporated from the solution which increases with increasing height is connected with the existence of dust particles in the air. Finally the authors thank A.N.Frumkin, Member, Academy of Sciences, USSR, for his mentioning the possibility of the discussed vertical distribution of H₂O₂. There are 2 figures and 7 references, which are Soviet.

ASSOCIATION: Odesskiy tekhnologicheskiy institut im. I.V. Stalina

(Odessa Technological Institute imeni I.V. Stalin)

PRESENTED: January 2, 1958, by A.N. Frumkin, Member, Academy of

Sciences, USSR

SUBMITTED: November 21, 1957

Card 3/4

	a Solu					
		1. Hydrogen	peroxide	vapors		
			i ing Lington			
Card	4/4					

 entere de la compania	Determination of the photolay	f the energer. Ukr.	rgy of activati khim. zhur. 2	on duri	ng action of:459-461 !5	
	1. Odesskiy tek fiziki. (Hydrogen p		eskiy institut (Heat of activ	factors and		

F '59.	(Electron emission)	(MIHA 12:4)	
	화장이 보면 꼭 얼룩요		
			· · · · · · · · · · · · · · · · · ·
	시민 학생들의 경기 얼마 하다는 것		

roykh,	T.L.
	Emission of electrons and H ₂ O during the corrosion of metals [with summary in Emglish]. Zhur.fiz.khim. 33 no.2:288-293 F 159. (MIRA 12:4)
	1. Odesskiy tekhnologicheskiy institut im. I.V. Stalina. (Electron emission) (Corrosion and enticorrosives)

5(4)

AUTHORS:

Roykh, I. L., Yarpovetskiy, L. Ya., (Odessa)

SOV/74-28-2-3/5

TITLE:

Chemical Electron Emission (Khimicheskaya elektronnaya

emissiya)

PERIODICAL:

Uspekhi khimii, 1959, Vol 28, Nr 2, pp 168-188 (USSR)

ABSTRACT:

In the present paper the authors give a general view of the main results obtained by investigations of the chemical emission. Since this is a matter of single investigations, no final conclusions can be drawn as yet. The chemical emission was investigated by various authors in various ways: by means of the drop-weight method (Refs 18, 19-26, 28 et al) at low pressure (Refs 44, 45) by point counter tube (Ref 31) with cylindrical counters (Refs 32, 34-36, 39, 40) as well as with copper tubes. Recently chemical emission has been investigated by an electron multiplier tube (Refs 41-43). From the results obtained it may be seen that the chemical emission is primarily conditioned by electrons. For this reason the authors consider the whole emission flow to be an electron flux in order to simplify matters and use the term chemical electron emission. The investigations of the

Card 1/4

Chemical Electron Emission

SOY/74-28-2-3/5

emission dependence on reactive substances have shown that the emission of charged particles occurs during a chemical reaction on the metal surface. It takes place only in the presence of active gases (Refs 17, 23, 25-28). Its intensity is increased with increasing energy released during the reaction (Ref 18). It was found that the presence of gaseous oxygen and a metal surface free from oxide ars prerequisites for the emission (Refs 35-38). On comparing the emission intensities of various metals the authors came to the conclusion that the intensities correspond to the position of these metals in the periodic system (Ref 41). The emission of refined metals, which apparently are not in any connection with chemical processes, was investigated (Refs 31, 32, 41, 42, 48). The conclusions drawn by numerous authors from the time dependence of emission on the oxidation mechanism of metals in various stages appear to be premature. The investigations of temperature dependence do not yet permit any generalization (Refs 31, 33-37, 40, 44, 45, 48, 54, 58). The dependence of the emission flow in the electric field and the distribution of emitted electrons according to energies were investigated in (Refs 23-26, 28, 43, 55). On connecting

Card 2/4

Chemical Electron Emission

SOV/74-28-2-3/5

the inhibiting field to the emitter the distribution of electrons can be found according to energies. Richardsor determined the distribution functions of the electrons, which were emitted by the influence of numerous active gases on the alloy KoNa, according to energies. With respect to the number of investigations carried out in this field, his publications are the only ones. Denisov and Richardson recommended the emission mechanism in 1934 (Refs 24, 27). It is their theory which permits the electron emission of gases on metal during the chemosorption to be explained. Numerous authors have shown that during the chemical reaction also an emission of negative ions is to be observed. This may be explained by the ionization of gas molecules during their reflection from the metal surface. The ionization of molecules may take place only when their electron affinity is greater than the work function of metal. This condition was confirmed for alkaline-metal halogens (Ref 44). The chemosorption of active gas molecules leads to electron emission. The ionic emission is caused by molecules which were not adsorbed on the metal. For this reason the electron emission may be considered to be a direct result of the

Card 3/4

Chemical Electron Emission

SOY/74-28-2-3/5

chemical reaction. The emission of negative ions must be regarded as an attendant phenomenon. According to certain reasons the separation of $\rm H_2O_2$ and the chemical electron emission may be considered to be connected processes accompanying the oxidation of metals. Other ways of excelectron emission are here described briefly, which were investigated in the course of past 10 years: a) emission during phase conversions ("crystal emission"), b) emission during destruction and deformation ("triboemission"), c) after-emission and d) induced photoelectric effect. In conclusion it is stated that the investigation of various ways of emission is still in its initial stage in spite of the relatively large number of publications. There are 10 figures and 124 references, 12 of which are Soviet.

Card 4/4

5(4), 18(7)

Roykh, I. L.

SOV/76-33-2-8/45

AUTHOR:

TITLE:

The Emission of Electrons and H_2O_2 in the Corrosion of Metals (Emissiya elektronov i H2O2 pri korrozii metallov)

PERIODICAL:

Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 2,

pp 288 - 293 (USSR)

ABSTRACT:

Investigations on electron emission in various reactions have been carried out by Haber and Just (Gaber and Yust) (Ref 1), Richardson et al (Ref 2), B. Krasyuk and V. Kalinin (Ref 5), Kramer (Ref 9), Haxel et al (Gaksel!)(Ref 11), Seidl (Seydl)(Ref 14), and Lohff (Loof)(Ref 15), and several other authors (Refs 3,4,6-10,12,13). References 7 and 16 report the finding that there is no relationship between the emission of electrons and the photographic effect of metals, but several important factors were not taken into consideration. In the work reported in this paper the authors show the relationship between the emission of electrons and H202

Card 1/3

in metal oxidation. The aluminum corrosion was investigated using a sharp point counter (Fig 1). The counter is a glass

The Emission of Electrons and H_2O_2 in the Corrosion of SOV/76-33-2-8/45 Metals

balloon with a plate serving as the cathode and a point fastened in the upper section of the balloon serving as the anode. The measured results (Fig 3) show that in the course of the five-hour measurement the emission as a function of the time is represented by a quadratic parabolic function. Photographic methods were developed to study the separation of H_2O_2 in the exidation of aluminum. For this purpose photographic film with a GOST sensitivity of -1.4 and a photo emulsion Nr 584 was used. Spectrally pure aluminum put at the disposal by Z. V. Vasil'yev was used. The same results were obtained by both the photographic methods used (one method is described in a special paper and another in reference 21). It was found

that in the 6 hour interval the $\rm H_2O_2$ separated according to the parabolic law. It is assumed that at least under certain conditions there exists a relationship between the electron emission and the separation of $\rm H_2O_2$ in metal

corrosion. There are 5 figures, 1 table, and 27 references, 8 of which are Soviet.

Card 2/3

The Emission of Electrons and H_2O_2 in the Corrosion SOV/76-33-2-8/45of Metals

ASSOCIATION: Odesskiy tekhnologicheskiy institut im. I. V. Stalina (Odessa Technological Institute imeni I. V. Stalin)

July 1, 1957 SUBMITTED:

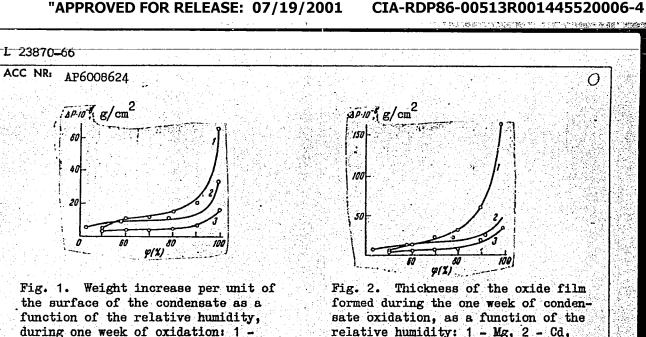
Card 3/3

ROYKH, I.L.; ÆLITSKAYA, S.G.; BOLOTICH, I.P.; ORDYNSKAYA, V.V.; NEDZVEDSKAYA, N.A.

Study of silicon exidation in air by optical polarization and photographic methods. Zhur. fiz. khim. 39 no.9:2306-2308 S 165. (MIRA 18:10)

1. Odesskiy tekhnologicheskiy institut imeni M.V. Lomonosova.

EWT(m)/EWP(t) SOURCE CODE: UR/0365/65/001/006/0677/0680 ACC NR: AP6008624 AUTHORS: Pustotina, S. R.; Tolkachev, V. Ye.; Rafalovich, D. M.; Roykh, ORG: Odessa Technological Institute im. M. V. Lomonosov (Odesskiy tekhnologicheskiy institut) PITLE: Oxidation of Mg, Zn, and Cd films formed by vacuum condensation in a humid 27 コン Zashchita metallov, v. 1, no. 6, 1965, 677-680 SOURCE: TOPIC TAGS: protective coating, metal film, corrosion resistance, magnesium, zinc, cadmium, metal oxidation ABSTRACT: Oxidation of Mg, Zn, and Cd vacuum condensates has been studied at various values of relative humidity. The information is of interest because the quality of the metallic films obtained by vacuum spraying is determined mainly by their atmospheric corrosion stability. The investigation was performed by gravimetric and polarized light methods, varying the relative humidity from 0 to 99%, at a temperature of 20C. The results of the study are summarized in Figs. 1 and 2. It was established that the increase of weight and the thickness of the oxidized layer are 4 and 2 times greater for Mg and Cd, respectively, than for Zn. At a relative humidity < 80% for Cd and Zn and < 70% for Mg, the protective oxide films are formed in 1 to 2 days of oxidation. The corrosion rate for all 3 condensates increases rapidly at relative



during one week of oxidation: 1 -Mg, 2 - Cd, 3 - Zn.

relative humidity: 1 - Mg, 2 - Cd, 3 - Zn.

humidity, r, above 90%. At these values of r, a porous layer with a large surface area is formed on the metal. Orig. art. has: 5 figures.

SUB CODE: 07, 11/ SUBM DATE: 01Mar65/ ORIG REF: 003/ OTH REF: 010

Card 2/2dla

L 16996-66 EMF()/EWT(m)/T IJF(c) RM/WW

ACC NR: AF6027287 (A) SOURCE CODE: UR/0191/66/000/008/0072/0073

AUTHOR: Kononchik, Ye. T.; Rafalovich, D. M.; Roykh, I. L.

ORG: none

TITLE: Oxidation of polymers in air during mechanical degradation

SOURCE: Plasticheskiye massy, no. 8, 1966, 72-73

TOPIC TAGS: peroxide, polyethylene, polystyrene, polycaprolactam, polymer degradation

ABSTRACT: The mechanical degradation of polymers may cause chemical reactions which evolve volatile substances, in particular, peroxy compounds. A photographic method was used to study the amount of volatile substances evolved during the mechanical degradation of low-pressure polyethylene, polystyrene, polycaprolactam and vulcanized rubber in air. The substances evolved caused a darkening on a photographic plate when it came in contact with its emulsion, and the degree of darkening was proportional to the amount of the substance. The composition of the volatile substances was identified by means of chemical indicators commonly employed for 1202 and by a luminescent method (luminol). The liberated organic peroxides (tert-butyl peroxy-acetate, tert-butyl peroxybenzoate, caproic peroxide, tert-butyl hydroperoxide and cumene hydroperoxide) had the same effect on the chemical and luminescent indicators as did 1202 and, like the latter, darkened the photographic plate. Teflon samples

Card 1/2

UDC: 678.019.31 : [678.742.2+678.746.22+678.675 126+678.44

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PUSTOTINA, S.R., TOLKSCHEV, V.Ye., RAFALOVICH, D.M., ROTKH, 1.1.

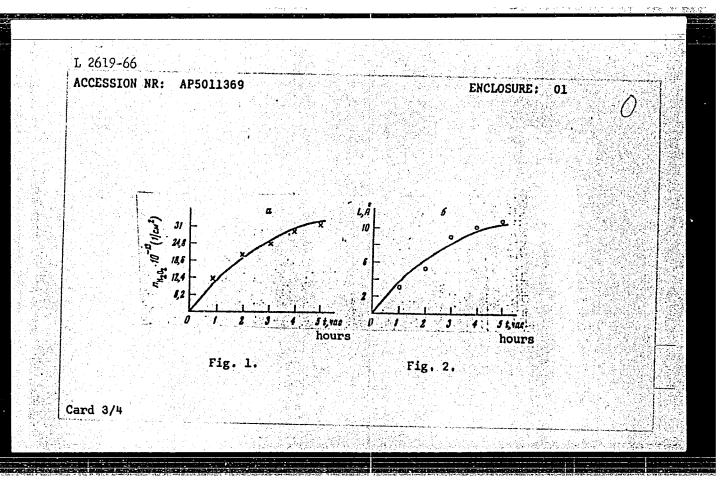
Oxidation of Yacuum Mg, En, and Cd condensates in a hunder atmosphere. Maskehamet. 1 no.62577-580 N-D 765.

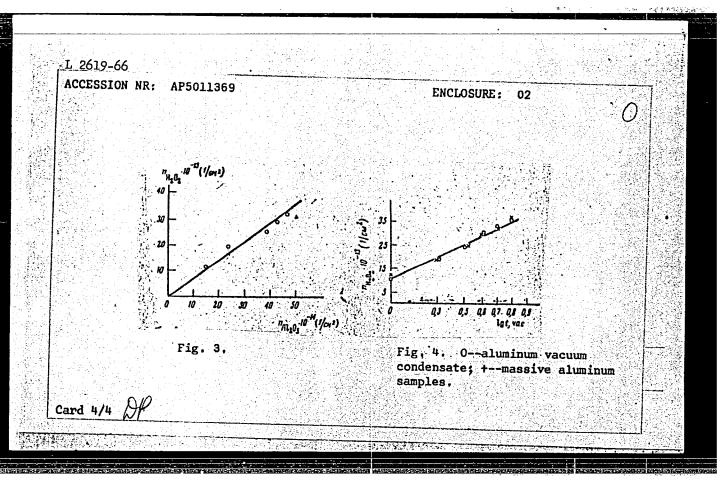
(NIPA 18610)

1. Odssakiy tekhnologicheskiy institat imeni M.V.lomonosova.

EVT(m)/EPF(c)/EWP(1)/EWP(t)/EWP(b) IJP(c) L 2619-66 JD/WB ACCESSION NR: AP5011369 UR/0365/65/001/002/0239/0241 620,193.2 Roykh, I. L.; Yefimovich, Ye. V.; Bolotich, I. P. 44.55 TITLE: On atmospheric corrosion of vacuum condensates of aluminum 44.55 B SOURCE: Zashchita metallov, v. 1, no. 2, 1965, 239-241 TOPIC TAGS: metal vapor deposition, vapor plating, corrosion resistance ABSTRACT: Atmospheric corrosion of vacuum condensates of aluminum was studied to examine the corrosion resistance of aluminum platings prepared by vacuum condensation, a technique widely used on a commercial scale. The samples, 500-5000 Å in thickness, were prepared by vacuum spraying of aluminum onto a glass gase. The extent of corrosion was measured by photographic and optical polarization techniques. The samples were oxidized for 10 min in air at 20 ± 2°C and at relative humidity of 50 ± 5%. In order to enhance the optical density, the aluminum films stretched on plates were immersed in a 4% Na2CO3 solution, and, then, immersed for 1 min in a 50% solution of ethyl alcohol and dried for 10 min at 100°C. The dependence of the number of evolved H2O2 molecules upon corrosion duration is shown Card 1/4

L 2619-66 ACCESSION NR: AP5011369 in fig. 1 of the Enclosure. The dependence of thickness of aluminum oxide layer (in Å) upon corrosion duration is shown in fig. 2 of the Enclosure. The dependence of the number of evolved H_2O_2 molecules upon the quantity of Al_2O_3 molecules formed is shown in fig. 3 of the Enclosure. The dependence of the number of evolved ${\rm H}_2{\rm O}_2$ molecules on the logarithm of corrosion time is shown in fig. 4 of the Enclosure. The correlation between the number of evolved H2O2 molecules and the number of Al₂0₃ molecules formed is: $n_{\text{Al}_2\text{O}_3} = 12 \cdot n_{\text{H}_2\text{O}_2}$. The linear dependence of the number of evolved $\mathrm{H}_2\mathrm{O}_2$ molecules upon the logarithm of corrosion duration is in agreement with data in the literature. Orig. art. has: 3 figures. ASSOCIATION: Odesskiy tekhnologicheskiy institut (Odessa Institute of Technology) SUBMITTED: 14Nov64 ENCL: 02 SUB CODE: MM, GC NO REF SOV: 003 OTHER: 003 Card 2/4





Decomposition of hydrogen peroxide vapors on the surface of patain and the role of Holy in atmospheric corrosion. Zhur. fts. knim. 38 nc.b:1586-1591 Je 162. (MIRA 18:3)

1. Odenskiy tekhnologichsskiy institut imeni Lomonosoya.

ACCESSION NR: AP5023694	UR/0076/65/039/009/2306/2308
44.5	541.17 YY.55 YY.55 YY.55
edzvedskaya, N. A. Y	YY, XY YY, XY G.; Bolotich, I. P.; Ordynskaya, V. V.;
ITIE: Study of the oxidation of sili hotographic method	$\frac{\cos n}{2}$ in air by the optical polarization and $\frac{n}{2}$
OURCE: Zhurnal fizicheskoy khimii, v	. 39, no. 9, 1965, 2306-2308
OPIC TAGS: silicon single crystal, h	ydrogen peroxide, oxidation kinetics
BSTRACT: The oxidation of the surface	e of an n-type silicon single crystal orient-
d in the Lill plane was studied at 70	0-73% humidity and 28-30°C. The kinetic re- of the oxide layer showed that this growth
beys the parabolic law $L^{***} = 54.3t$.	During the first three house following the
resuly cleaned silicon surface has an	nickness of 17.5 Å. It was found that the effect on a photographic film, and the pho-
opraphic density D was plotted as a se	mction of the exposure time. Chemical analy-

· 1143-66		
cordance between the kinetics evolution of H ₂ O ₂ indicates to dation of silicon in air. Exvapors of a 10% aqueous solut H ₂ O ₂ . Thus, the fraction of formed during the oxidation.	that the latter may serve , , speriments showed that the sution of hydrogen peroxide dec H ₂ O ₂ evolved amounts to only Orig. art. has: 2 figures.	the criterion for the oxi- prface of silicon under composes 96.2% of absorbed y a minute part of the H ₂ O ₂
		-V. Lomonosova (Odessa
ASSOCIATION: Odesskiy tekhno Technological Institute) /// SUBMITTED: 31Ju164		-V. Lomonosova (Odessa SUB CODE: GC
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ROYKH, I.L.; YEFIMOVICH, Ye.V.; BOLOTICH, 1.P.

Atmospheric corrosion of aluminum vacuum condensates. Zashch.
met. 1 no.2:239-241 Mr-Ap '65. (MIRA 18:6)

1. Odenskiy tekhnologicheskiy institut.

ROYKH, I.L. Determination of the absolute amount of H₂O₂ evolved in the atmospheric corrogion of metals. Zhur. fiz. khim. 36 no.62 1335-2337 Je¹62 (MIRA 17:7) 1. Odesskiy tekhnologicheskiy institut.

ROYKH, I.t., BOLGTICH, I.P.; KOLTUNOVA, L.N.

Determination of the sctivation energy of formation of hydrogen oxide and hydrogen peroxide in the atmospheric corrosion of Mg and Al. Zhur. fiz. khim. 36 no.3:2052-2054 S *62.

(NIRA 17:6)

i. Gdesskiy tekhnologicheckiy institut imeni Lomonosova.

ROYKH, L.I.; RAFALOVICH, D.M.

Relation between the weight increase of the oxide film and the amount of H₂O₂ evolved in the atmospheric corrosion of magnesium. Zhur, fiz. khim. 36 no.6:1198-1201 Je⁶62 (MIRA 17:7)

1. Odesskiy tekhnologicheskiy institut.

ROYKH, I.L.; BELITSKAYA, S.G.

Atmospheric corrosion of magnesium in the corona discharge.
Zhur.fiz.khim. 37 no.8:1694-1698 Ag '63. (MIRA 16:9)

1. Odesskiy tekhnologicheskiy institut.
(Magnesium--Corrosion) (Electric discharges)

ROYKH, I.L.; KONONCHIK, Ye.T.

Dependence of the amount of emitted H₂O₂ in the atmospheric corrosion of aluminum an air pressure. Zhur.fiz.khim. 37 no.2: 433-436 F ¹63. (MIRA 16:5)

1. Odesskiy tekhnologicheskiy institut imeni Ibmonosova.
(Aluminum--Corrosion) (Hydrogen peroxide)

ROYKH, I.L.; ORDYNSKAYA, V.V.; BOLOTICH, I.P.

Effect of mechanical treatment on the surface area of metals.

Dokl. AN SSSR 146 no.6:1316-1317 0 '62. (MIRA 15:10)

l. Odesskiy tekhnologicheskiy institut im. M.V. Lomonosova.
Predstavleno akademikom L.A. Artsimovichem.
(Surfaces (Technology)) (Metals—Finishing)

S/076/63/037/002/014/018 B144/B180

AUTHORS:

Roykh, I. L., Kononchik, Ye. T.

TITLE:

Dependence between the H202 volume evolved in atmospheric

corrosion of Al on the atmospheric pressure

PERIODICAL:

Zhurnal fizicheskoy khimii, v. 37, no. 2, 1963, 433-436

TEXT: The evolution of $\rm H_2O_2$ as a function of the partial oxygen pressure was studied in an apparatus where the elevated surface of an Al disk rotating at 1 rev/0.72 sec was dressed continuously by a fixed cutter and the $\rm H_2O_2$ molecules separating owing to corrosion were registered photographically. The pressure effect of the air was investigated in the 5-760 mm Hg range by placing the whole apparatus in a vacuum drum, evacuating for 3-6 min, drying with $\rm P_2O_5$ for 10 min, rotating the disk under the cutter for 3 min, and admitting air for 1 min. The film was sensitized with 4% $\rm Na_2CO_3$ and 50% ethanol, and by heating to $\rm \sim 90^{\circ}C$

Card 1/2

Dependence between the H202 ...

S/076/63/037/002/014/018 B144/B180

after exposure. D the optical density of the film blackening, increased sharply between 5 and 100 mm Hg and then fell steadily. Above 25 mm Hg the curve was parabolic. A diagram is given for converting D into the number of $\rm H_2O_2$ molecules evolved, the method will be described elsewhere. The curve shows that $\rm H_2O_2$ at first evolves at the expense of atmospheric oxygen and then a protective film forms. The curves plotted in the 0.06-0.4 sec range for the kinetics of $\rm H_2O_2$ evolution in atmospheric corrosion show that the process is linear up to 25 mm Hg and parabolic from 25 to 300 mm Hg. There are 8 figures.

ASSOCIATION:

Odesskiy tekhnologicheskiy institut im. M. V. Lomonosova

(Odessa Technological Institute imeni M. V. Lomonosov)

SUBMITTED:

July 24, 1961

Card 2/2

5/020/62/146/006/008/016 B104/B186

AUTHORS:

Roykh, I. L., Ordynskaya, V. V., Bolotich, I. P.

TITLE:

The influence of machining on the finish size of metal

surfaces

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 146, no. 6, 1962, 1316-1317

TEXT: The influence of different machining methods (cutting, shaping, milling and grinding) on the true surfaces of Mg, Al, steel Cr -3 (St-3), steel Cr -45 (St-45), bronze and cast iron is investigated using a profilometer of the type Kalibr-VEI. With this instrument, surfaces of the 6th and up to the 14th class of surface finish can be examined. The enlargement varied between the limits of $2 \cdot 10^3$ and $12 \cdot 10^4$ vertically, between 116.7 and 4200 horizontally. In the instrument a diamond tip (radius of curvature 1.25 μ) exerts a pressure of 0.1 g against the metal surface. For all metals and all grades of finish the ratio of $n = S_{\rm measured}/S_{\rm Gecm} = 1/\sin(\alpha/2)$ was almost equal to unity. The angle α , defined as the apex angle of the four-faced pyramids constituting the metal

Card 1/2

S/020/62/146/006/008/016 B104/B186

The influence of machining on the ...

surface, showed only small variations from $168 \pm 2^{\circ}$ in all of the test pieces. From the results of 200 profilograms it follows that the kind of machining and the degree of surface finish exert little influence on Smeasured. Differences between true and measured surface values are attributed to unevennesses characteristic of surface qualities far exceeding the highest measurable classes of finish quality. There is 1 figure.

ASSOCIATION: Odesskiy tekhnologicheskiy institut im. M. V. Lomonosova

(Odessa Technological Institute imeni M. V. Lomonosov)

PRESENTED: May 28, 1962, by L. A. Artsimovich, Academician

SUBMITTED: May 25, 1962

Card 2/2

ROYKH, I.L.; ORDYNSKAYA, V.V.

Effect of the mechanical treatment of the surface on the evolution of H₂O₂ in the atmospheric corrosion of metals. Zhur.prikl.khim. 34 no.9:1979-1986 S '61. (MIRA 14:9)

1. Odesskiy tekhnologicheskiy institut imeni I.V. Stalina. (Metals—Corrosion) (Hydrogen peroxide)

27342 s/080/61/034/009/005/016 D204/D305

18.83 00

Roykh, I.L., and Ordynskaya, V.V.

Influence of mechanical surface treatment on H202 AUTHORS:

evolution in the atmospheric corrosion of metal. TITLE:

PERIODICAL: Zhurnal prikladnoy khimii, v. 34, no. 9, 1961.

1979 - 1986

TEXT: The influence on the mechanism of oxidation of magnesium. aluminum, zinc and cadmium of abrading their surfaces with wheats of defferent grain sizes was investigated by means of a photographic method. This was based on the photographic activity of H202 evolved during the corrosion of the above metals. For investigating the H_2O_2 evolution kinetics during oxidation, six metal eposimens, cleaned by hand on an abrasive wheel, were placed on a protographic plate for various lengths of time. Successive specimens were removed from the plate at two minute intervals. The photogra-

Card 1/4

27342 S/080/61/034/009/005/01c D204/D305

Influence of mechanical surface ...

phic plates were developed 15-20 hours after exposure. Six degrees of optical blackening ($D_{\rm kin}$), depending on the time of contact between specimen and photographic emulsion, were obtained on each plate. The values of $D_{\rm kin}$ were used in order to plot H_2O_2 evolution kinetics curves, n-t, where n is the relative number of H_2O_2 particles having struck the photographic plate. The value of n was found from the characteristic curve representing the dependence of D on lg t. In order to obtain the characteristic curve, aluminum specimens, cleaned on an emery wheel, no. 180, were placed on a photographic plate for 2, 4 ... 24 minutes, the specimen surface being cleaned every two minutes. This enabled the quality of H_2O_2 molecules separated from the metal and hitting the photographic emulsion film, to be increased to an integer number. The number of molecules separated in two minutes was taken as the unit of measurement. The magnitude of the values of $D_{\rm char}$. obtained by photometering, was found to depend on the length of contact between the Card 2/4

27342 S/080/61/034/009/005/016 D204/D305

Influence of mechanical surface ...

specimen and the photographic plate. It was found that in the first stages of atmospheric corrosion of metals, the quantity of H202 evolved increased with increase of surface roughness. Kinesic investigations of H₂O₂ evolution in an interval of from 2-12 minutes and from 1-5 hours, after grinding the surface with emery wheels of six degrees of coarseness, showed that a parapolic relationship exists between the rate of H202 evolution and the caree of surface coarseness. On flat grinding of metals to different degrees of surface cleanliness, the increase in roughness, as detramined by the degree of unevenness of the profile, affects metals of high chemical activity to a greater extent, increasing their rate of corrosion. Passivation by preliminary radiation in a corona discharge field considerably decreases the influence of roughness. Similar results are obtained in the later stages of canation (12 minutes), as in the earlier ones (2 minutes), owing to the formation of the protective film. In the interval of 1-5 hours after the beginning of oxidation, a change in roughness has practi-Card 3/4

27.34.2
Influence of mechanical surface ... S/080/61/034/009/005/016

cally no effect on H₂O₂ evolution. There are 7 figures, 3 tables and 12 references: 11 Soviet-bloc and 1 non-Soviet-bloc. The reference to the English-language publication reads as follows: P.M. Aziz and H.P. Godard, J. Electrochem. Soc., 104, 738, 1957.

ASSOCIATION: Odesskiy tekhnologicheskiy institut imeni I.V. Stalina (Odessa Technological Institute imeni I.V. Stalin)

SUBMITTED: August 11, 1960

Card 4/4

1454, 1521, 1496, 4016

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AUTHORS:

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Car

Roykh, I. L., Rafalovich, D. M. The effect of relative humidity of the air on the formation of H_2O_2 in atmospheric corrosion of magnesium and aluminum

TITLE:

Zhurnal prikladnov khimii, v. 34, no. 4, 1961, 864 - 870

By means of a photographic method the evolution of H2O2 during the first 15 minutes and the first 6 hours of atmospheric corrosion of magnesium and aluminum, the relative air humidity warying from 0 to 90 %, was investigated. PERIODICAL: It was observed that the amount p of H₂O₂ formed increases with ϕ , and log p = 0.00 h. To the first 6 hours of correction at ϕ = 0.00 d the avaluation of It was observed that the amount p of hour formed increases with T, and log p a ϕ b. In the first 6 hours of corrosion at ϕ = 0 - 60 % the evolution of H₂O₂ is varying. (A) parabolic dependence between p and time t is observed at $\phi = 60$ %, while for $\phi \leq 60$ % and the intervals 0 - 3 hours and 3 - 6 hours parabolic equations with different coefficients are valid. The effect of relaparabolic equations with different coefficients are valid. The effect of relative humidity () of the air was investigated gravimetrically by W. H. I. Vernon (Ref. 1: Trans. Farad. Soc., 27, 255, 582, 1931; and Ref. 2: Trans. Farad. Soc., (Ref. 1: Trans. Farad. Soc., 27, 255, 582, 1931; Lukonina et al. (Ref. 4: ZL, 1668, 1935) on steel and copper, and by T.I. Lukonina et al. (Ref. 4: ZL, 1668, 1935) on 516 (D. 16) and DOS (W. OS) aluminum allows. 22, 1463, 1955) on A 16 (D 16) and B95 (V 95) aluminum alloys. A sharp increase

Card 1/6

The effect of relative humidity

26865 \$/080/61/034/004/006/012 A057/A129

on the bottom of the cylindrical hermetically closed glass box, where the experiments were carried out. After exposure to the ${\rm H}_2{\rm O}_2$ evolved by the sample at a certain humidity in the test box, the photoplates were developed and the optical density Do of the blackening was determined. The dependence of the optical density D of the photoplate blackening after exposure to a solution of ${\rm H}_2{\rm O}_2$ of a certain concentration at a certain humidity was determined and corresponding curves were plotted. From these curves and values obtained with metals the dependence of the evolved H2O2 amount p on was estimated (Figure 4). The observed increase in p with is in agreement with literature data (Ref. 1,2,4) indicating an increase in the oxide film with increasing | . In the present investigations also the amount p of H2O2 evolved from the metals during corrosion at varying was determined and the results are shown in Figure 5, demonstrating that for ψ = 0 - 90%, $\log p = a^{\circ} + b$ (where a and b are different for the interval 0 - 30%and for 30 - 90 %). These results are in agreement with data given by N., D. Tomashov and A. A. Lokotilov (Ref. 15: Sb. "Korroziya i zashchita staley" ("Corrosion and protection of steel" Mashgizdat 158, 1959). Kinetics of H202 evolution were studied during the first 6 hours of corrosion for % = 0,15, 30, 45 and 60 % The amount of H202 formed during the first 15 minutes was considered as unit in these experiments. The obtained results plotted in squares of the formed H2O2 Card 3/6

The effect of relative humidity

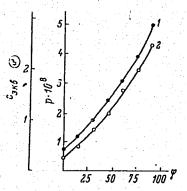
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amount versus corrosion time are shown in Figures 8, 9. For f=60% the parabolic equation $p^2=kt$ (2) is valid while for f=60% the function shows two segments. Approximately for 0 - 3 hours of corrosion equation (2), and for 3 - 6 hours equation $p^2=k_1t+k_2$ (3) is valid. There are 9 figures and 15 references: 12 Soviet-bloc and 3 non-Soviet-bloc.

SUBMITTED: August 1, 1960

Figure 4: Dependence of the amount (p \cdot 10^8 , g/cm²) and concentration (c_{equiv} %) of hydrogen peroxide evolved from the metals during the first 15 minutes of oxidation on relative humidity \cdot (%). 1 - magnesium, 2 - aluminum, 3 - c_{equiv}.

Card 4/6



ROYKH, I. L., Doc Chem Sc.1, "Separation of H_2O_2 in The Atmospheric corrosion of metals." Moscow, 1960. (State Com of Council of Ministers USSR for Chemistry, Order of Labor Red Banner Sci Res Phys-Chem Inst im L. Ya. Karpov). (KL, 2-61, 200).

2

Relation between the H.O. evolved and the quantity of the oxide nolecules formed in the atmospheric corresion of Mg and Al. Dekh. Am SSSR 137 no. 1:125-129 Mr-Ap '51. (MHA 14:2)

1. Cdosskiy tekhnologichaskiy institut im. T.V. Stellina.

Prodstavleno akaconikon A.M. Frunkinya.

(Aluminum-Gerrection) (Lagresium-Gerresion)

(Nydrogen peromide)

5/020/61/137/001/018/021 B101/B204 and Bolotich, I. P. Roykh, I. L. AUTHORS: 10 Selection duragen requireded light and the quentity of the כענונונ oxide molecules formed in the atmospheric corrosion of Mg and 15 Doklady Akademii nauk SSSR, v. 137, no. 1, 1961, 126-129 PERIODICAL: TEXT: It is the purpose of the present paper to prove that the photographic method of studying the atmospheric corrosion of metals by the action of the $\mathrm{H}_2\mathrm{O}_2$ formed upon the photoemulsion is physically justified. 30 For this purpose, a comparison between the results obtained by the photographic method and those obtained by studying thin oxide films in polarized light was carried out in the same way as done by P. Drude (Ref. 3), L. Tronstad, and B. Winterbottom (Ref. 4). For the parameters Δ and ψ of the elliptic polarization of light, the Eq. (1) is written 25 down. Card 1/6

35

Relation between separated... S/020/61/137/001/018/021 B101/B204

$$L = \frac{\Delta - \overline{\Delta}}{A \left(1 - \frac{1}{n_1^2}\right)}; \quad n_1^2 = \frac{1 + \frac{2\psi - 2\overline{\psi}}{\Delta - \overline{\Delta}} \cdot \frac{\cos^2 \varphi - a}{a' \sin 2\overline{\psi}}}{\cos^2 \varphi}, \tag{1}$$

$$A = -\frac{4\pi}{\lambda} \frac{\cos \varphi \sin^2 \varphi (\cos^2 \varphi - a)}{(\cos^2 \varphi - a)^2 + a^2}; \quad a = \frac{1 - x^2}{n^2 (1 + x^2)^2}, \quad a' = \frac{2x}{n^2 (1 + x^2)^2}$$

L = layer thickness, n_1 = refractive index of the layer, λ = wavelength of the incident monochromatic light, n = refractive index of the metal, κ = absorption coefficient of the metal, φ = angle of incidence of the beam, ψ = angle between the oscillation plane of the incident polarized light and the plane of incidence. In the case of reflection, not only the amplitudes of the components, but also the phases change: δ_{ii} = δ_{ii} = Δ . Δ and ψ relate to the pure metal surface, Δ and ψ to the oxidized surface. For the optical constants of the pure metal surface it holds that:

n = $\sin\varphi \tan\varphi \cos 2\sqrt{(1 + \cos\Delta \sin 2\sqrt{})}$; $\kappa = \sin\Delta \tan 2\sqrt{}$ (2). In order to

Card 2/6

Relation between separated...

S/020/61/137/001/018/021 B101/B204

create unique reflection conditions, magnesium and aluminum were polished by means of pastes of the GOI (State Institute of Optics) until the unevennesses no longer exceeded 0.14 - 0.1 μ . By means of a polarization-goniometer constructed by V. A. Marchenko, $\overline{\Delta}$ and $2\overline{\psi}$ were first determined, and herefrom n and κ were calculated. Next, Δ and 2ψ were determined in the course of the oxidation process, and herefrom L and n_1 were calculated from (1). ACBA-120A (SVD-120A) lamp, brightness about 5000 stilb, was used for illumination. The following was found with λ = 5890 A: ϕ = 60°, for aluminum n = 2.37, X = 1.53, n_1 = 1.51; for magnesium n = C.484, X = 6.93, n_1 = 1.7. As shown in Fig. 1, there is good agreement between the photographic and optical data of the oxidation kinetics. For the increase of L, L² = 31t was found for aluminum; for magnesium L².69 = 63t was found. The photographic method was checked by determining the dependence of the density of blackening on the concentration of H_2O_2 vapor. Experiments were carried out at $(20\pm1)^{\circ}$ C, the distance between photographic plate and H_2O_2 solution amounted to 2 mm. The

Card 3/6

Relation between separated ...

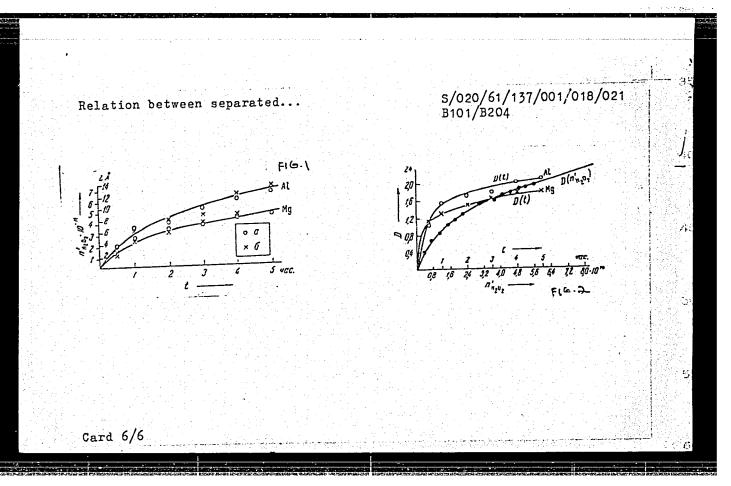
S/020/61/137/001/018/021 B101/B204

volume v of the ${\rm H_2O_2}$ solution evaporated within one minute was found to be 0.218 mg/cm²min for ${\rm H_2O_2}$ concentrations between 0.05 - 2 %. According to G. Skatchard (Ref. 7), the molar fraction ${\rm y_h}$ of the ${\rm H_2O_2}$ vapor over the ${\rm H_2O_2}$ solution was calculated, and from ${\rm n_{H_2O_2}^{\dagger}} = {\rm vy_h^{N/M}_h}$ (3)

(N = Avogadro number, M_h = molecular weight of H_2O_2), the number n' of the evaporated H_2O_2 molecules was calculated. The effect produced by the distance between plate and surface of the solution was taken into account by $n_{H_2O_2}^! = n_{H_2O_2}^0 \exp(-0.417 \cdot 2)$ (4) and, accordingly, the curves

D=D(n') were drawn (Fig. 2). Agreement among the values obtained by means of the photographic and optical method respectively proves that between H_2O_2 separation and thickness of the oxide layer there exists linear dependence. This interrelation was more closely studied in consideration of the true surface of the metal. Investigation of the polished Card 4/6

Relation bet	ween separated	S/020/61/137/001/018/021 B101/B204
(All-Union F 0.01 - 2 \mu c geometrical unevenness c lowing was f	ause an increase of t surface by a maximum coefficient 2.5 determ cound; 1H ₂ O ₂ 11.51	ilometer of the type "Kalibr" - VEI ute) showed that unevennesses of he true surface as compared with the of only 3%. In consideration of the nined by 0. Erbacher (Ref. 9), the foliated of the sined by 0. Erbacher (Ref. 9), the foliated of the sined by 0.
from the cor	clusion was drawn tha	it a considerable part of H2O2 disintegrates
again on the	metal surface. The	re are 2 figures and 9 references:
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APPROVED FOR RELEASE: 07/19/2001 CIA-RDP86-00513R001445520006-4"

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ASD/ESD-3/AFWL/IJP(C)/SSD Pz-L/Pab-L/PI-L JD/AT
ACCESSION NR: AP3004968 S/0076/63/037/008/1694/1698 \$5

AUTHOR: Roykh, K. L.; Belitskaya, S. G.

TITLE: Atmospheric corrosion of magnesium in corona discharge

SOURCE: Zhurnal fiz. khimii, v. 37, no. 8, 1963, 1694-1698

TOPIC TAGS: corrosion, protective film, corona discharge, oxidation of metal, inhibition of corrosion, Mg, Zn, Cd, Al, Fe, Cu

ABSTRACT: Authors analyzed the conditions and rate of a protective film on magnesium which is caused by irradiation of corona discharge. A magnesium cylinder, 14 mm in diameter and 15 mm high containing 0.004% Fe,0.009% Si and 0.0021% Mn, was polished and placed in a corona field 20 mm away from the electrode and then irradiated from 15 to 90 seconds. Corona discharge was produced by an induction coil. Potential difference was 12 kV. Degree of oxidation was determined by photographic method based on the activity of separated hydrogen peroxide.

Card 1/2

L 18315-63

ACCESSION NR: AP3004968

Findings indicate that: (i) irradiation in corona field produces an intensive protective film; (ii) effective distance of sample from the electrode is 20-25 mm; (iii) oxidation rate constant is in hyperbolic dependence upon time of primary action of discharge; (iv) irradiation lowers the rate of subsequent oxidation of magnesium in atmosphere; (v) preliminary attempts with Cd and Zn show that protective film formation also takes place on these metals. Orig. art. has: 7 figures.

ASSOCIATION: Odesskiy tekhnologicheskiy institut (Odessa

technological institute)

SUBMITTED: 30Sep60

DATE ACQ: 06Sep63

ENCL: 00

SUB CODE: PH. CH

NO REF SOV: 011

OTHER: 003

Card

2/2

GURVICH, E.B.; ROYKHEL', V.M.

Epidemiology of vater-borne typhoid fever. Zhur.mikrobiol.epid. i immun. 28 no.5:30-33 My '57. (MLRA 10:7)

1. Iz Sanitarno-epidemiologicheskoy stantsii g.Voskresenska Moskovskoy oblasti.
(TYPHOID FEVER, epidemiol.
water-borne outbreak)

ZETTELENOK, N. A.; REYZIN, F. N.; ROYKHEL',V..M.; GOL'DFARS, M. M.

"Fiziologiye vzaimodeystviya esno-virusov s biologicneskimi substratami, fiziologicheskaya rol' sn-grupp virusov."

report presented at Symp on Virus Diseases, Moscow, 5-9 Oct 64.

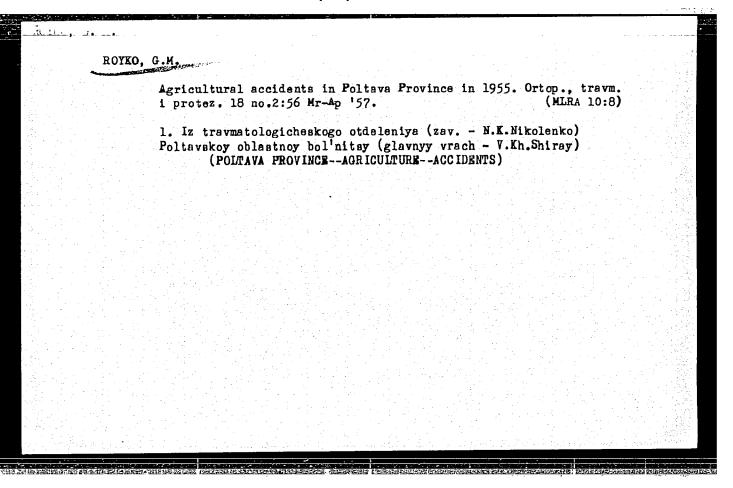
Institut poliomielita i virusnykh entsefalitov AMN SSSR, Moskva.

	Purification and concentration of a polyvolent and	i-int wenze	
	therepoutte and prophylactic horse secum. Vop.med. 98 163.	virus no.8:92- (MIRN 17:10)	
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ROYKIEWICA, Andrzej

Scientific problems of psychic hygiene; in connection with the 6th Congress of the Polish Association of Psychic Eygiene in Marsaw, May 9-12, 1963. Nauka polska 11 no.5:167-173 '63.

1. Polska Akademia Mauk, Zaklad Migieny Psychicznej i Psychiatrii Dzieciecej, Marszawa.



ROYMER, V.A. KERNEYCHUK, G.P.

Method of approximation for determining the macrostructure of porous catalysts. Zhur.fiz.khim.28 no.10:1812-1819 0'54. (MIRA 8:2)

1. Akademiya nauk USSR. Institut fizicheskoy khimdi im. L.V. Pisarzhevskogo, Kiyev. (Catalysts)

Drying grain khoz. 21 no	in by the use of so. 3:23-25 '63.	sots. sel'- (MIRA 16:8)	
l. Tekhnich Respublika	neskiy universitet, (GreinDrying)	Germanskaya Demokratic (Sorbents)	heskaya

BARABADZE, I.I.; BAKRADZE, G.S.; BERIDZE, G.I.; VAKHVAKHISHVILI, N.I.;
GABUNIYA, G.A.; GABUNIYA, Sh.V.; GANGIYA, A.A.; GOGOBERIDZE, Ya.A.;
DZIMISTARISHVILI, A.I. [deceased]; ZNAMENSKIY, K.F.; KVANTALIANI,
N.A.; NIKOLAYSHVILI, V.S.; TOPADZE, L.I.; KHUNTSARIYA, A.G.; YAKO-BASHVILI, N.Z.; DZHOMARDZHIDZE, G.S., red.; ROYNISHVILI, N.I., red.;
PRITYKINA, L.A., red.; KISINA, Ye.I., tekhn. red.

[Food industry of the Georgian S.S.R. during the last 40 years]
Pishchevaia promyshlennost Gruzinskoi SSR za 40 let. Moskva,
Pishchepromizdat, 1961, 162 p. (MIRA 14:9)
(Georgia-Food industry)

ROYNISHVILI. N. M.

"Special Calculations in the Protection of a Railroad Bed Against Landslide Phenomena." Dr Tech Sci, All-Union Sci-Res Inst of Railroad Transport, Min Railways, Moscow, 1954. (KL, No 1, Jan 55)

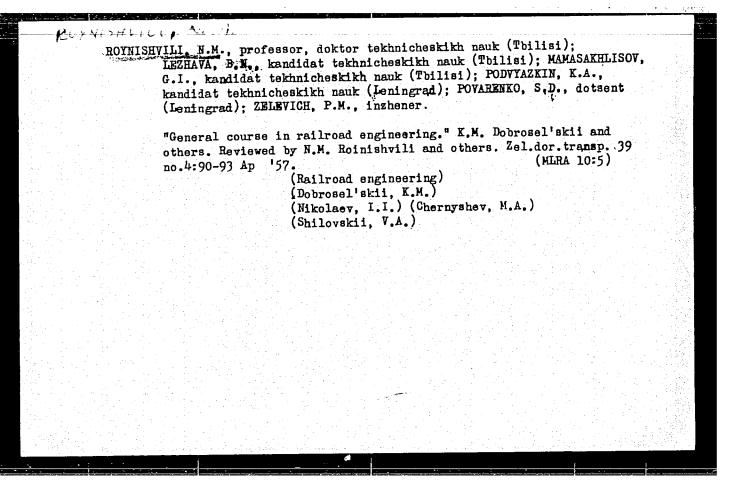
Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (12)

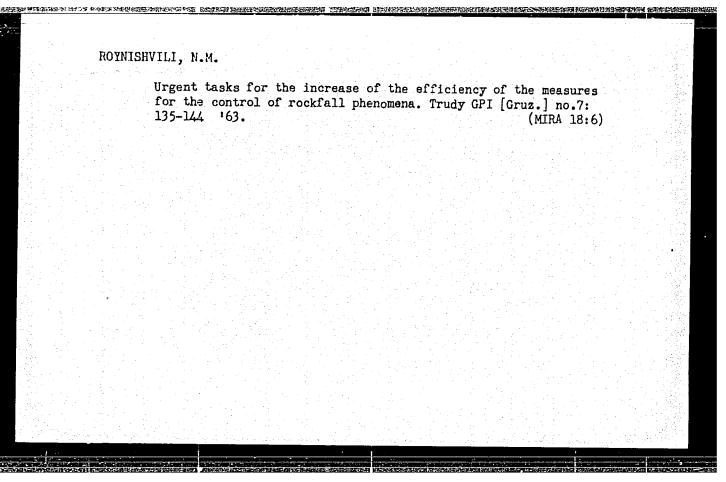
so: SUM No. 556, 24 Jun 55

ROYNISHVILI, N.M., detsent, kandidat tekhnicheskikh nauk.

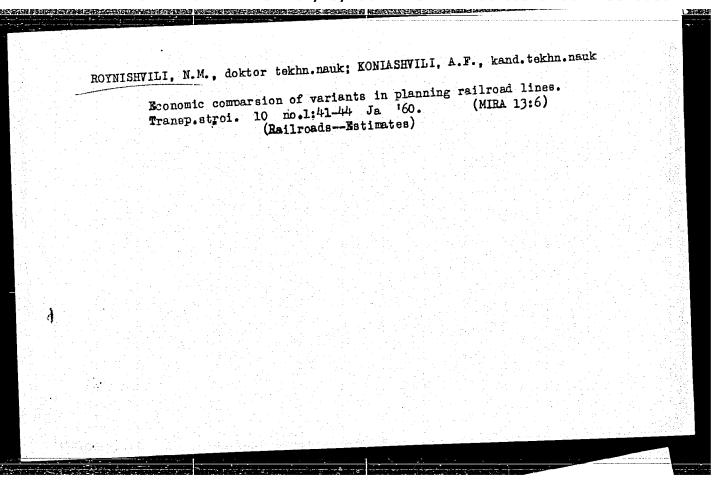
Initial data and metheds of determining the basic dimensions of landslide protection structures. Trudy Tb IIZHT ne.28:35-104 '55. (Railread engineering) (Landslides) (MIRA 9:2)

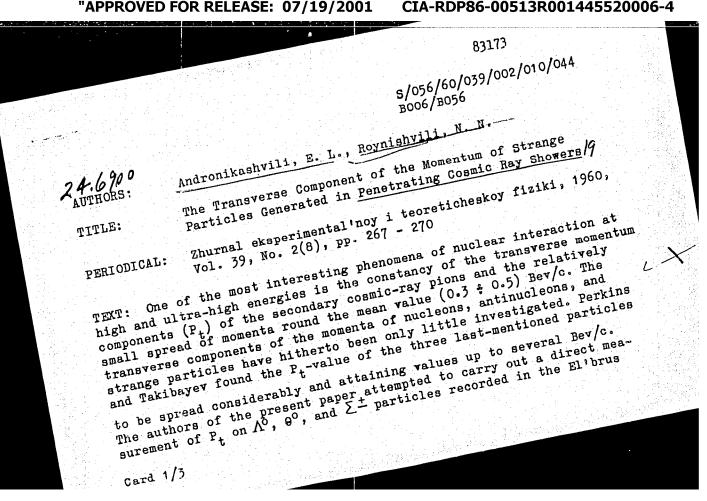
	Anti-landslide	catch ditches.	Transp.stroi. 6	no.5:15-17 My (MLRA	'56. 9:8)
		(Mountain railr	oads)		
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preventive	structures. T	proving methods of deransp. stroi. 13 no.5	:47-50 My 163.	My 163.	
	(Landslides)	(Mountain railroads)	(MIRA 16:7)		





83173 s/056/60/039/002/010/044 B006/B056

The Transverse Component of the Momentum of Strange Particles Generated in Penetrating Cosmic Ray Showers

laboratory. The particles were recorded by means of a double cloud chamber in a magnetic field. The apparatus and method are more closely described in Refs. 2-7. Measurements were carried out on $20\,\Lambda^{\circ}$, $21\,9^{\circ}$, and $30\,\Sigma^{\pm}$ particles, whose other parameters were known. In order to and $30\,\Sigma^{\pm}$ particles, whose other parameters were known. In order to characterize the recording probabilities of the particles, each decay event was provided with a weight W_1^{-1} , where W_1 is the a priori probability of its recording. The error in the number of events recorded per bility of its recording. The error in the number of events recorded per P_t interval is then given by $(\Sigma\,W_1^{-2})^{1/2}$. The recording probability \overline{W} as a function of P_t is shown in Fig. 1. Up to about 2 Bev/c, \overline{W} shows no spread. Fig. 2 shows the P_t -distribution of all decay events of strange particles corrected in this way. The mean values P_t measured are given in a table. The P_t -spectrum has its most probable value in the momentum range (0.2 - 0.4) Bev/c, and attains values of up to 3 Bev/c. It is,

Card 2/3

ROYMISHVILI, Nikolay Minovich, prof., doktor tekhn.nauk; DOBSHITS, M.L., inzh., red.; BCBROVA, Ye.N., tekhn.red.

[Structures to protect railroads from earth slides] Protivoobval'nye sooruzheniia na zheleznykh dorogakh. Moskva, Vses. izdatel'sko-poligr.ob"edinenie M-va putei soobshcheniia. 1960.

227 p. (Railroad engineering)

CIA-RDP86-00513R001445520006-4 "APPROVED FOR RELEASE: 07/19/2001

SOV/124-58-3-3283

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 3, p 107 (USSR)

Roynishvili, N. M. AUTHOR:

Initial Data and Methods of Determining the Basic Dimensions TITLE:

of Protective Structures Against Slides (Iskhodnyye dannyye i metody opredeleniya osnovnykh razmerov protivoobval'nykh

zashchitnykh sooruzheniy)

PERIODICAL: Tr. Tbilis. in-ta inzh. zh. -d. transp., 1955, Nr 28,

pp 35-104

Observational results are given describing the fall of rocks ABSTRACT:

on 17 open-cut sections of railway track differing in type of profile, slope grade, rock-falling height, character of surface covering, geological structure, and type of rock formations. Fall of rocks was caused by works necessitating climbing on top of the open cuts. Motion of a rock within the limits of a single rebound consists in the trajectory equation of a body thrown at a certain angle to the surface of the slope with an initial velocity v; the velocity vector forms an angle desig-

nated β with the vertical. The magnitude of β was determined indirectly according to the greatest distance of rock

Card 1/4

SOV/124-58-3-3283

Initial Data and Methods of Determining the Basic Dimensions (cont.)

fall l. Data processing of the in-situ observations was performed max statistically by plotting the nomograms of relative recurrence of velocity values as dependent upon the height of fall H and the determination of the parameters of the probability-distribution curve. Several curves for the values of rock velocities are given. For calculation purposes, the final velocity of rocks was determined by the semiempirical formula

$$v_k = \mu \sqrt{2gH}$$
 $(\mu = \sqrt{1 - k \cot \alpha})$ (1)

where a is the embankment slope angle and k is a coefficient characterizing the resistance to the motion of rocks along the slope. Curves of the calculated velocities and coefficient values for μ and k are given for the velocity range of up to 40-45 m/sec. Depending on the steepness of the slope three velocity ranges are defined: 1) decelerating motion on gently sloping faces (less than 28°), where a fall starts at some initial velocity and decelerates gradually to zero; 2) accelerating motion on steep slopes (from 28° to 60°) where a fall starts with a rolling motion and develops into rebounding motion; and 3) free fall on extremely steep slopes (over 60°). For cases of tree and shrubcovered slopes a coefficient $\eta < 1$ is introduced into formula (1). Expressions for v_k are given for cases of irregular slopes consisting of sections Card 2/4

SOV/124-58-3-3283

Initial Data and Methods of Determining the Basic Dimensions (cont.)

of various grades. A velocity-vector angle formula is selected

$$\beta = [200 + 2 \alpha (1 - \alpha^{\circ} / 45^{\circ})] v_{k}^{-1/3}$$

Determination of β allows the use of conventional formulas for the rockrebound trajectory elements and the determination of the overall dimensions of necessary protective structures [G.M.Shakhunyants. Zemlyannoye polotno zheleznykh dorog (Earthen Railroad Track Beds), Moscow, Transzheldorizdat, 1953]. Observational data as to the greatest height of rock flight over the head of the rails h_{max} are given and the recovery coefficients are calculated according to the formula $\rho = \sqrt{\frac{h \max}{H + (v_k^2 / 2 \text{ g}) \cos^2 \beta}}$

$$\rho = \sqrt{\frac{\frac{h \max}{h \max}}{H + (v_k^2 / 2 g) \cos^2 \beta}}$$

Calculation of the overall dimensions for lateral antislide-protection structures of railway tracks consists in the calculation of rock flight and rebound over reach and are based on finding the rock trajectory for the most unfavorable conditions possible in the vicinity of the protective structure. Design calculation data for strength and stability of protective antislide structures are given. An approximate method of dynamic design calculation of Card 3/4

SOV/124-58-3-3283

Initial Data and Methods of Determining the Basic Dimensions (cont.)

protective structures equipped with an earth fill for the purpose of damping the impacts is given, and a recommended procedure for the calculation of such structures is demonstrated. Bibliography: 10 references.

G. I. Ter-Stepanyan

Card 4/4

MANDZHAVIDZE, Z.Sh.; ROYNISHVILL, N.N.; GERSAMIYA, D.V.; KOZLOV, A.A.;
KOTLYAREVSKIY, D.M.; PURTSELADZE, T.D.; TATALASHVILI, N.G.;
SHTEMANETYAN, G.Z.

Lifetime of charge ∑ thyperons. Trudy Inst.fiz.AN Gruz.SSR
8:125-129 '62. (Hyperons)

(Hyperons)

KOZLOV, A.A.; KOTLTAREVSKIY, D.I.; ROYNISHVILI, M.N.; TATALASHVILI, M.G.;
TSAGARRELI, E.I.; TSINTSBADZE, A.I.; TSINTSADZE, V.D.; DZIDZIGURI,
R.I.

Method of studying tracks in the Wilson magnetic chamber. Soob.
AN Gruz. SSR 19 no.2:143-150 Ag '57. (MIRA 11:3)

1. Institut fiziki AN GruzSSR, Tbilisi. Predstavleno akademikom
E.L. Andronikashvili. (Gloud chamber)

AUTHORS:

Mandzhavidze, Z. Sh., Roynishvili, N. N., SOV/56-34-5-9/61

Chikovani, G. Ye.

TITLE:

The Coservation of the Decays of Charged Particles in a Double Cloud Chamber (Nablyudeniye raspadov zaryazhennykh chastits v

sdvoyennoy kamere Vil'sona)

PERIODICAL:

Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1958,

Vol. 34, Nr. 5, pp. 1110-1115 (USSR)

ABSTRACT:

This paper analyzes !O decays of heavy charged particles. These particles were observed by means of a device which is similar to the device of C.H. Jork et al. (Ref 2). Investigations were carried out in the Vysokogornaya El'brusskaya kosmiches-kaya laboratoriya (El'brus High Mountain Cosmic Laboratory). For the measurements discussed in this paper a rectangular double cloud chamber was used. This cloud chamber consists of two independent volumes (each of them has the dimensions 280x100x110 mm) and three sections for the absorber. The two independent volumes are united by the same carcass. The cloud chamber was filled with argon (1000 torr) and a mixture of 70% ethyl alcohol and 30% water was used as condensate. The

Card 1/4

70 % ethyl alcohol and 30 % water was used as condensate. The first series of experiments was carried out with copper ab-

The Observation of the Decays of Charged Particles in a Double Cloud Chamber

sov/56-34-5-9/61

sorbers, the second with lead absorbers. The magnetic field strength in the working volume had the value 4300 Oe. 11559 photographs were taken within 2836 hours, and 2269 penetrating showers were recorded by these photographs. Moreover. 10 forked tracks were observed on these photographs, they may be interpreted as V-decays. The authors found also 22 / decays, 1 decay of a timeson, 1 decay of 2pions (while they were moving) and 13 stars. The results of the measurements of the momenta, angles and the approximate values of the ionization are compiled in a table. All the observed decays, (with the exception of one), taking account of the observation errors, lie within the allowed range for hyperons and K-mesons. Only one case can be exactly interpreted as the decay of a K-meson, for all the other cases it is impossible to discern between K - and Y - decays. Among the decay products no proton was found. The V^{\pm} -decays, are divided into two groups, according to the character of production. The 6 particles of the first group have a very low ionization caused by the primary particles. The second group consists of 4 slow particles with rather a high ionization. These 4 particles are not con-

Card 2/4

The Observation of the Decays of Charged Particles in a Double Cloud Chamber

sov/56-34-5-9/61

nected with a visible interaction and are generated far from the place of the decay. One decay is interpreted as the decay of a particle which is heavier than a K-meson. It is possible to assume that this particle is the charged analogon of the neutral meson the decay of which was observed by Kovan (Ref 12). The authors thank Professor E.L. Andronikashvili for supervising these investigations, and also the collaborators of the Tbilisskiy gosudarstvennyy universitet (Tbilisi State University), L.D. Gedevanishvili and E.I. Tsagareli, and also the collaborators of the Institut fiziki (Physics Institute) R.I. Dzidziguri, A.I. Tsintsabadze, V.D. Tsintsadze. There are 4 figures. 3 tables, and 13 references, 5 of which are Soviet.

ASSOCIATION: Institut fiziki Akademii nauk Gruzinskoy SSR (Physics Institute AS Georgien SSR) Tbilisskiy gosudarstvennyy universitet (Tbilisi State University)

SUBMITTED:

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Card 3,

CIA-RDP86-00513R001445520006-4 "APPROVED FOR RELEASE: 07/19/2001

\$/627/60/002/000/027/027 D299/D304

3-24/0(2205,2705,1559)

Mandzhavidze, Z. Sh., Roynishvili, N. N., Chukovani, G. Ye., Kozlov, A. A., Kotlyarevskiy, D. M., Tatalash-AUTHORS:

vili, N. G., and Tsintsibadze, A. I.

Study of penetrating showers at an altitude of 2000 m TITLE:

above sea level

International Conference on Cosmic Radiation. Moscow, SOURCE:

1959. Trudy. v. 2. Shirokiye atmosfernyye livni i kas-

kadnyye protsessy, 338-341

The properties of unstable heavy particles were studied by means of a magnetic cloud chamber with lead absorbers. Among 8700 nuclear interactions, 139 cases of decay of neutral particles were observed, as well as 29 decay processes of charged strange particles. In addition, 11 decay processes, described by the authors in an earlier work, are also included in the study. As a result of the investigation of neutral particles, 45 V°-shaped tracks were iden-Card 1/4

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Study of penetrating ...

tified as decays of \bigwedge^{o} -hyperons, and 38 - as θ^{o} -mesons. Fifty-six of the remaining V^0 -shaped tracks could not be identified. Out of 40 V-particles, I was interpreted as T-meson decay, 7 could be interpreted as K-meson decay and 2 - as Σ -hyperons. The other particles could not be interpreted by decay-dynamics only; for their interpretation considerations had to be employed which proceed from the considerable difference in the lifetime of hyperons and K-mesons respectively. In Solov yev's work (Ref. 3: preprint O.I.Ya. I.) it is shown that for strong interactions involving strange particles, there are no obvious theoretical assumptions which would require conservation of parity. If such interactions are not invariant with respect to space inversion, one should expect the appearance of hyperon polarization in the plane of generation. These considerations were used as a basis for constructing the angular distribution protons of the decay of $\Delta^{\rm o}$ -particles with momenta below 800 MeV./c. Further, the authors investigated the lifetime of

Card 2/4

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Study of penetrating ...

 \triangle particles by 2 methods. By the first method, they obtained for the mean lifetime the value

$$T_{\Delta 0} = (2,83 + 2,32) \cdot 10^{-10} \text{ sec}$$

The second method yielded

$$T_{\Delta 0} = (3.02 + 1.14) \cdot 10^{-10} \text{ sec}$$

Further, an attempt was made to determine the lifetime of \leq -hyperons. Earlier results in this respect are in disagreement. It was found that 13 of the decay processes of charged particles can be considered as ϵ^{\pm} -hyperons. The lifetime of 9 of these particles is

Card 3/4

Study of penetrating ...

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$$\tau_{\Sigma \pm} = \langle (0,57 \pm 0,36) \cdot 10^{-10} \text{ sec} \rangle$$

There are 1 table and 9 references: 3 Soviet-bloc and 6 non-Soviet-bloc. The references to the English-language publications read as follows: S. Hayakawa. Phys. Rev., 108, 1533, 1957; D. A. Glaser. Ann. International Conference on High Energy Physics at CERN, 1958; I. Snayder, W. Y. Chang and I. G. Gupta. Phys. Rev., 106, 149, 1957.

ASSOCIATION: Institut fiziki AN Gruz.SSR (Physics Institute AS Georgian SSR)

Card 4/4

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Chikovani, G. Ye.
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AUTHORS:

' Mandzhavidze, Z. Sh., Roynishvili, N. N.

TITLE:

Strange-particle energy distribution

PERIODICAL: Referativnyy zhurnal, Fizika, no. 3, 1962, 50, abstract 3B412 ("Pizikis institutis shromebi. Sakartvelos SSR Metsnierebata

Akademia, Tr. In-ta fiz. AN GruzSSR", 1960, v. 7, 197 - 200, English

summary)

Spectra of \bigwedge^0 , Θ^0 , and \sum^{\pm} particles generated in lead by cosmic ray particles with a mean energy of ~20 Bev were taken at 1,800 m above sea level using a rectangular double Wilson chamber placed in the magnetic field. The spectra of Λ^{0} particles and Σ^{\pm} hyperons have similar distributions of about equal width with a sharp maximum in the 500-Mev region, while the $\theta^{\rm O}$ meson spectrum appears as sloping and spreads up to 5.5-Bev energies. This result may be explained either by the strong anisotropy of the angular distribution of Λ^0 and Σ^{\pm} hyperons in the center-of-inertia system or by similar features of the generation of different types of baryons on quasi-free nucleons of the target. [Abstracter's note: Complete translation]

Card 1/1

ANDRONIKASHVILI, E.L.; ROYNISHVILI, N.N.

Transverse component of the pulse of strange particles generated in penetrating showers of cosmic rays. Zhur. eksp. i teor. fiz. 39 no.2:267-270 Ag 160. (MIRA 13:9)

1. Institut fiziki Akademii nauk Gruzinskoy SSR. (Cosmic rays)

Analysis of the distrange particles	istribution of . Zhur.eksp.i	transverse im teor.fiz. 41	pulses of p no.3:919-	ions and 923 S '61. (MIRA 14:10)
l. Institut fizik	i AN Gruzinsko Particles (Nuc	y SSR. lear physics))		

S/048/62/026/006/004/020 B125/B112

AUTHORS: Mandzhavidze, Z. Sh., and Roynishvili, N. N.

TITLE: Some problems of strange particle production

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Seriya fizicheskaya, v. 26, no. 6, 1962, 716 - 721

TEXT: A study was made of the energy distribution of the heavy unstable particles, with energies of some dozens of Bev, present in the penetrating cosmic radiation showers. Further, the distribution of the transverse momenta of the strange particles was investigated on the basis of observations made on $50\Lambda^{\circ}$, 42 0°-, and $18\Sigma^{\pm}$ -particles in a Wilson chamber at 1800 m above sea level. The obvious similarity of the Σ^{\pm} - and Λ° -hyperons points to a similarity of the baryon energy distribution in multiple points to a similarity of the baryon energy distribution in cascade particle production at some dozens of Bev. If the intranuclear cascade particle production at some dozens of Λ° -, Λ° - and Λ° -particles equally, processes influence the production of Λ° -, Λ° - and also of the Λ° -hyperons is much then the energy distribution of the Λ° - and also of the Σ^{\pm} -hyperons is much

Card 1/3

S/048/62/026/006/004/020 B125/B112

Some problems of strange particle production

more anisotropic than that of the θ^0 -mesons. The present results are well consistent with new results got with 8 Bev pions in a propane chamber of the OIYaI. The difference in the mean values of the transverse momenta p_1 of the non-pion particles is probably due to the small increase of p_1 in the energy range between the threshold and 10 Bev. If the thermal motion of the liquid elements causes the p_1 -spectra, then the p_1 -spectrum of the mixture of the particles observed determines their temperature of departure $kT(1.1^{+0.6}_{-0.1})m_{\pi}c^2$. This temperature agrees well with the transverse pion momenta.

 $\left\langle \left(\frac{P_{\perp}}{mc}\right)^{2}\right\rangle = \left\langle \left(\frac{P_{\perp}}{mc}\right)^{2}\right\rangle_{a=0} + a^{2}\left(1 + 2\left\langle \left(\frac{P_{\perp}}{mc}\right)^{2}\right\rangle_{a=0}\right),$

are the root mean square momenta as a function of the hydrodynamic velocity. At different values of a, the root mean square momenta depend approximately linearly on their temperature of departure. If Landau's theory is valid also if $E_0^2 = 10^{12}$ ev, then the transverse heavy particle momenta will probably increase rapidly if the energy increases from 10^{10} to 10^{12} ev. If

Card 2/3

s/048/62/026/006/004/020 Some problems of strange particle production

the conditions of existence of the transverse momenta imply Heisenberg's uncertainty principle, then the results of the present paper indicate the structure of the region of particle production. The region of strange particle production has the radius $\langle r^2 \rangle_V^{1/2} > 0.37 \cdot 10^{-13}$ cm. There are 8 figures. The most important English-language reference is: E. R. Awu-nor-Renner, L. Blaskovith, R. French, C. Chesquer, Y. B. de Minvielle, Devau, W. W. Neale, C. Pelletier, P. Rivet, A. B. Sahiar, Y. O. Skillicon,

Card 3/3

Nuovo cimento, 17, 134 (1960).

RAZDOL'SKAYA, L.A.; ROYNISHVILI, N.N.; GABUNIYA, L.L.; MANDRITSKAYA, K.V.; TATALASHVILI, N.G.

Program for processing the tracks of penetrating cosmic ray showers with energies of 10^{10} to 10^{12} ev. Fiz. chast. vys. (MIRA 18:12) energ. no.1:65-84 '65.

EWT(d)/EWT(m)/FCC/T ACC NR: AP5024662 SOURCE CODE: UR/0048/65/029/009/1784/1787 Roynishvili, N.N.; Mandritskaya, K.V. ORG: none TITLE: Analysis of experimental data characterized by power law distributions /Report, All-Union Conference on Cosmic Ray Physics held at Apatity 24-31 August 1964/5 SOURCE: AN SSSR. Izvestiya. Şeriya fizicheskaya, v. 29, no. 9, 1965, 1784-1787 16,44,54 TOPIC TAGS: statistics, error statistics, distribution function, nuclear physics, cosmic ray ABSTRACT: This paper is concerned with the determination of the exponent m in a distribution of the type xmdx from experimental data. The frequently employed method of least squares for estimating m and the chi-square criterion for comparing hypothetical values of m are not only very laborious for a power law distribution but they do not give optimum results. The maximum likelihood estimate of m and the likelihood ratio criterion for comparing hypothetical values of m are presented and their advantages are pointed out. In order to normalize the distribution the range of the independent variable x is restricted to the interval (0, 1). The likelihood function then becomes a very simple function of the average of the logarithms of the sample (experimental) values and the probability distribution for the likelihood ratio becomes an incomplete Card 1/2

gamma function, tables of which a nate for the power law distributi least square estimate, it is effi is pointed out and removed. The with several examples from the fithank Z.Sh.Mandzhavidze and G.Ye. the results. Torig. art. has: 16	on is not only ver cient. The blas o application of the elds of nuclear an Chikovani for valu	y simple to comput f the maximum like methods discussed d cosmic ray physi able advice during	e but, unlike the lihood estimate is illustrated cs.// The authors
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Analysis of IAd. fiz. 1	experimental da no.6:1028-1031	ta satisfying type Je '65.	x ^m dx. distributions. (MIRA: 18:6)	
1. Institut	fiziki AN Gruzi	nskoy SSR.		